

SYNTHESIS OF DIPHOSPHIRENES BY [2+1]-CYCLOADDITION OF IMINO-PHOSPHANES
 TO A PHOSPHAALKYNE AND ISOMERIZATION TO
 AZADIPHOSPHETINES

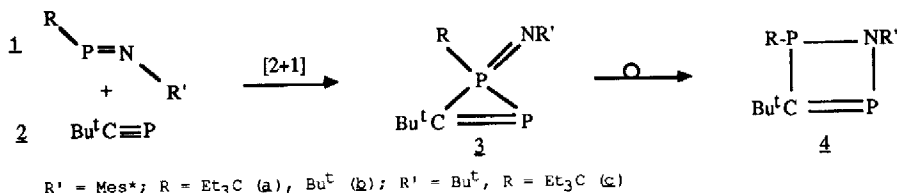
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Summary: The reaction of iminophosphanes, $RP=NR'$ **1** with $P=Cu^t$ **2** affords di-
 phosphirenes $R(R'N)=P-P=Cu^t$ **3**, and by isomerization the azadiphosphetines
 $RP-N(R)-P=Cu^t$ **4**.

The (2+1)-cycloaddition of phosphalkynes with carbenes^[1] or the heavier carbene ana-
 logues^[2,3] is an elegant route to the three-membered ring systems containing the phos-
 phaalkene-moiety ($-P=CR-$).

As we have demonstrated on the basis of spectroscopic and experimental results imino-
 phosphanes containing the C-P=N - skeleton are exceptional candidates for carbenic
 reaction behaviour^[4]. Here we report the synthesis of a novel three-membered ring
 system the diphosphirene **3** and its isomerization to the 1,2,4-azadiphosphetine-2 **4**.



Treatment of the $P=Cu^t$ ^[5] (5 mmol) **2** with one equivalent of the iminophosphanes
 (**1a**^[4b,6], **1b**^[7]), at -30°C in *n*-hexane (10 ml) produces a cleanly deep red solution of
 the [2+1]-cycloaddimers **3a,b**, which can be separated as a deep red solid by evaporation of
 the solvent. The compositions of the thermal unstable air- and moisture sensitive di-
 phosphirenes **3a,b** have been confirmed by elemental analysis and mass spectrum. The mass
 spectrum shows the molecular ion peaks with low intensity (70 eV, $m^+ m/z$ 489 **3a**, 448 **3b**
 (1%). The loss of R and PCu^t from this ion appears to be a strongly favoured process
 which gives the base peak (PNMes^+ m/z 290). The constitution of **3a,b** has been proven by
 means of their $^{31}\text{P}\{^1\text{H}\}$ n.m.r. spectra. Indicative for the two-fold coordinated phospho-
 rus atom is the low field shift (δ 351.0 ppm **3a**, 349.5 **3b**) while the coupling constants

of 182 Hz **3a**, 180 Hz **3b** are in good agreement with a one bond PP-interaction ($^1J_{PP}$ 140 - 230 Hz in azadiphosphiridines [**4b**]). While **3a**^[8] remained unchanged in solution at 25°C, **3b** changed within hours (within few minutes at 50°C) under clear up in colour and formation of a new product **4b**^[8], which possesses the same molecular ion peak in the mass spectrum as **3b** but a quite different fragmentation pattern (70eV, m/z 448 (m^+ 0.1%), 246 (Mes⁺ 5%), 202 (m^+ - Mes⁺ 18%), 57 (Bu⁺ 100%). Furthermore in the $^{31}P\{^1H\}$ -n.m.r. spectrum the highly shielded phosphorus nucleus in **3b** (δ - 41.5 (-43.2 **3a**)) is shifted drastically to lower fields (δ 160.8), while the signal of the two coordinated phosphorus atom remains unchanged (δ 330.1). This together with the decrease in J_{PP} (68.5 Hz) is consistent with the formation of a 1,2,4-azadiphosphetene-2 the thermodynamically favoured product of the cycloaddition reaction.

The same ring system **4c**^[8] has been formed by reaction of **2** with the sterically less hindered iminophosphane **1c**, without observation of the intermediate **3c**. Isomerization of kinetically formed [n+1]-cycloadducts of iminophosphanes to the thermodynamically favoured ones has been observed recently in the case of a 1,2 λ^3 ,3 λ^5 -azadiphosphiridine^[9] (the [2+1]-self-addition product of "carbenic" iminophosphane). Four-membered ring systems containing the -P=C< moiety are known so far as phosphacyclobutenes^[10] and 1,3-diphosphacyclobutenes^[11].

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References and Notes:

- [1] O.Wagner, G.Maas and M.Regitz, *Angew.Chem.,Int.Ed.Engl.*, 26 (1987) 1257.
- [2] A.Schäfer, M.Weidenbruch, W.Saak and S.Pohl, *Angew.Chem.,Int.Ed.Engl.*, 26 (1987) 776.
- [3] A.H.Cowley, S.W.Hall, C.M.Nunn and J.M.Power, *J.Chem.Soc.,Chem.Commun.* 1988, 753.
- [4] a) E.Niecke, D.Gudat, W.W.Schoeller and P.Rademacher, *J.Chem.Soc.,Chem.Commun.* 1985, 1050; b) E.Niecke, M.Lysek, E.Symalla, *Chimia* **40** (1986) 202; c) E.Niecke and M.Lysek, *Tetrahedron Lett.* **29** (1988) 605.
- [5] G.Becker, G.Gressner, V.Uhl, *Z. Naturforsch., Teil B*, **36** (1981) 16.
- [6] L.N.Markovski, A.B.Drapailo, V.D.Romanenko, A.V.Ruban, *Zh. Obshch. Khim.* **56** (1986) 714.
- [7] D.Barion, Diplomarbeit, Bonn 1987.
- [8] **3a**: deep red crystals; m.p. 102°C (dec.). **4b**: light yellow crystals; m.p. 190-192°C. **4c**: orange oil; b.p. 68°C, 0.01 Torr, MS, m/z 301 (m^+ 2%), 147 (Bu⁺CPNP⁺ 100%). ^{31}P -NMR(C_6D_6): δ 126.0, 342.2 ($^2J_{PP}$ 57 Hz); ^{13}C -NMR(C_6D_6): δ 218.0 (d, d 31.2, 33.5 Hz) >P=C=P-.
- [9] E.Niecke, D.Gudat, and E.Symalla, *Angew.Chem.,Int.Ed.Engl.* **25** (1986) 834.
- [10] J.Fink, W.Rösch, U.-J.Vogelbacher, and M.Regitz, *Angew.Chem.,Int.Ed.Engl.* **25** (1986) 280.
- [11] R.Appel, V.Barth and F.Knoch, *Chem.Ber.* **116** (1983) 938.

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